First Synthesis of the 3'-Sulfated Lewis^a Trisaccharide, Putative Ligand for the Leucocyte Homing Receptor

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The trisaccharide 3'-OSO₃Na-Gal $\beta(1 \rightarrow 3)$ [Fuc $\alpha(1 \rightarrow 4)$]GlcNAc (3'-sulfated Le^a) is prepared from 4-methoxybenzyl β -D-*N*-acetylglucosaminide.

The L-selectin (homing receptor), an adhesion molecule on leucocytes, plays a key role in lymphocyte extravasation into peripheral lymph nodes1 and neutrophile recruitement at inflammatory sites.² It has been shown that this adhesion protein recognizes 3'-sialyl Lewisa (SLea) and 3'-sialyl Lewisx (SLe^x),³ but Feizi et al. recently⁴ found that the 3'-sulfated analogues of SLe^a and SLe^x are even better ligands. However, in these crucial studies, an equimolecular inseparable mixture of 3'-sulfated Lea and 3'-sulfated Lex tetrasaccharides derived from an ovarian cystadenoma glycoprotein was used, which precluded determination of the most potent of the two sulfated oligosaccharides as an L-selectin ligand. In order to answer this question, we decided to chemically synthesize the terminal 3'-sulfated Lea trisaccharide on a preparative scale. It must be emphasized that structures of all of the ligands of the three selectins (E, L and P) have not yet been determined and that chemical synthesis will surely help to solve these questions and provide carbohydrates with adverse inflammatory inhibition properties.

The new 4'-methoxy benzyl glycoside 1[†] was readily prepared from the common O, N-peracetylated glucosaminyl chloride in 96% yield using classical conditions [Hg(CN)₂, toluene]. The use of the 4-methoxybenzyl group at the reducing position relies on the possibility to remove it selectively, if necessary, to further extend the oligosaccharide structure. After de-O-acetylation (NEt₃-MeOH-H₂O, 1:8:1), reaction with benzaldehyde dimethyl acetal in tetrahydrofuran (THF) in the presence of TsOH afforded the acetal 3^{\dagger} (93%) which was glycosylated⁵ with acetobromo-

$$R^{3}O_{R^{1}O_{R^{1}O_{R^{1}O_{R^{2}}}} OMBn$$

$$1; R^{1} = R^{2} = R^{3} = Ac$$

$$2; R^{1} = R^{2} = R^{3} = H$$

$$3; R^{1} = H, R^{2}, R^{3} = CHPh$$

$$R^{2}O_{R^{1}O_{R^{2}O_{R^{2}}}} OMBn$$

$$A; R^{1} = R^{2} = Ac, R^{3}, R^{4} = CHPh$$

$$b; R^{1} = R^{2} = H, R^{3}, R^{4} = CHPh$$

$$b; R^{1} = All, R^{2} = H, R^{3}, R^{4} = CHPh$$

$$b; R^{1} = All, R^{2} = Ac, R^{3}, R^{4} = CHPh$$

$$b; R^{1} = All, R^{2} = Bn, R^{3} = H, R^{4} = Bn$$

$$OR^{2}O_{R^{2}O_{R^{2}O_{R^{2}}}} OR^{2}O_{R^{2}O_{R^{2}}} OR^{2}O_{R^{2}O_{R^{2}O_{R^{2}}}} OR^{2}O_{R^{2}O_{R^{2}}} OR^{2}O_{R^{2}O_{R^{2}}}} OR^{2}O_{R^{2}O_{R^{2}}} OR^{2}O_{R^{2}O_{R^{2}}} OR^{2}O_{R^{2}O_{R^{2}}}} OR^{2}O_{R^{2}O_{R^{2}}} OR^{2}O_{R^{2}O_{R^{2}}}} OR^{2}O_{R^{2}O_{R^{2}}}} OR^{2}O_{R^{2}O_{R^{2}}} OR^{2}O_{R^{2}O_{R^{2}}}} OR^{2}O_{R^{2}O_{R^{2}}} OR^{2}O_{R^{2}}} OR^{2}O_{R^{2}O_{R^{2}}}} OR^{2}O_{R^{2}O_{R^{2}}} OR^{2}O_{R^{2}}} OR^{2}O_{R^{2}O_{R^{2}}} OR^{2}O_{R^{2}}} OR^{2}O_{R^{2}}} OR^{2}O_{R^{2}} OR^{2}O_{R^{2}}} OR^{2}O_{R^{2}}} OR^{2}O_{R^{2}} OR^{2}O_{R^{2}}} OR^{2}O_{R^{2}}} OR^{2}O_{R^{2}}} OR^{2}O_{R^{2}} OR^{2}O_{R^{2}}} OR$$

13; $R^1 = SO_3Na$, $R^2 = R^3 = H$

[†] All new compounds gave satisfactory elemental analysis. Physical data are given below:

data are given below: **i**: m.p. 164–165 °C, $[\alpha]_{D}^{20}$ -49 (c 0.9 CH₂Cl₂), **2**: m.p. 210–211 °C, $[\alpha]_{D}^{20}$ -41 (c 1, H₂O). **3**: m.p. 285 °C, $[\alpha]_{D}^{20}$ -54 (c 1, pyridine). **4**: m.p. 136–137 °C, $[\alpha]_{D}^{20}$ -39 (c 1.07, CH₂Cl₂). **5** m.p. 251 °C, $[\alpha]_{D}^{20}$ -67 (c 1, DMF). **6**: m.p. 247 °C, $[\alpha]_{D}^{20}$ -74 (c 0.5, DMF). 7 m.p. 175 °C, $[\alpha]_{D}^{20}$ -14 (c 1.5, CHCl₃). **8**: m.p. 137 °C, $[\alpha]_{D}^{20}$ -34 (c 1.5, CHCl₃). 9: m.p. 167 °C, $[\alpha]_{D}^{20}$ -11 (c 1, CHCl₃). **10**: $[\alpha]_{D}^{20}$ -40 (c 1.5, CHCl₃). 11: $[\alpha]_{D}^{20}$ -54 (c 1, CH₂Cl₂). **12**: $[\alpha]_{D}^{20}$ -42 (c 1, CH₂Cl₂). **13**: as a mixture of α and β anomers ($\alpha'\beta$, 1.5 : 1), $[\alpha]_{D}^{20}$ -38 (c 0.5, MeOH); ¹H NMR (250 MHz, D₂O) δ 5.11 (H-1 α), 5.02 (H-1"), 4.86 (H-5"), 4.71 (H-1 β), 4.62 (H-1' α), 4.58 (H-1' β), 4.29 (H-3', H-4'), 3.86 (H-2 α), 3.79 (H-2", H-4"), 3.76 (H-4 α), 3.74 (H-4 β , H-6 $\alpha\beta$, H6b β), 3.61 (H-5 β), 3.61 (H-2), 3.55 (H-5'), 2.05 (CH₃CONH), 1.18 (CH₃ Fuc).

galactose [Hg(CN)₂, molecular sieves (4 Å), nitromethanetoluene, 1:1]. The blocked disacccharide obtained in quantitative yield was then treated with NEt₃ in aqueous methanol $(NEt_3-MeOH-H_2O, 1:8:1)$ for 48 h at room temp. to give 5⁺ (93%). The 3'-position was then protected with an allyl group through the stannylene procedure⁶ (1-Bu₂SnO, toluene; 2-AllBr, BrNBu₄, toluene) to give the crystalline compound 6† in 77% yield purified as the peracetylated derivative 7.† After de-O-acetylation (NEt₃-MeOH-H₂O, 1:8:1, for 60 h at 80 °C), perbenzylation is performed in dimethylformamide (DMF) with benzyl bromide (4 equiv. per OH) in the presence of HNa (1.1 equiv. per OH) added portionwise during 5 h to avoid N-benzylation. In these conditions, compound 8^{\dagger} was obtained in 65% yield (along with a mixture of dibenzylated derivatives which is recycled to furnish finally compound 8 in 78% yields). Regioselective opening of the benzylidene acetal using sodium cyanoborohydride-HCl(g) in THF,7 gave the compound 9[†] in 81% yield. Then, the fucose residue was introduced from freshly prepared perbenzyl fucosyl bromide8 under in situ anomerization9 conditions using tetraethylammonium bromide in DMF-CH₂Cl₂ in the presence of diisopropylethylamine. The fully protected tetrasaccharide 10[†] was thus obtained in 80% yield. Two-step deallylation [1-(Ph₃P)₃RhCl, 2-HgCl₂-HgO, acetone-water] delivered compound 11[†] in 67% yield (90% based on starting material recovery). Then sulfation with the sulfur trioxide-trimethylamine complex in DMF (12 h at 55 °C) gave the sulfated trisaccharide 12[†] (82%). Complete deprotection [10% Pd/C, H_2 (1 atm)] afforded in 80% yield the sodium salt of 3'-sulfated Lewis^a trisaccharide 13 as an amorphous white powder after purification by silica gel chromatography (PrⁱOH-AcOEt-H₂O, 3:5:2) followed by cation exchange chromatography (AG50W-X8, Na⁺) and lyophilisation of the

aqueous solution. Compound 13, obtained as a mixture of α and β anomers (α/β , 1.5:1) gave satisfactory elemental analysis and has been fully characterized by NMR using a COSY (correlation spectroscopy) experiment.† In particular, the H-3' is found at $\delta 0.7$ downfield compared with H-3' of the Lea trisaccharide10 showing unambiguously the presence of the sulfate on galactose at O-3. We have thus prepared 60 mg of the title compound, the biological properties of which are currently being studied.

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